[V(3rd Sm.)-Biochemistry-G/SEC-A-1/CBCS]

2021

BIOCHEMISTRY — GENERAL

Paper : SEC-A-1

(Tools and Techniques in Biochemistry)

Full Marks : 80

The figures in the margin indicate full marks. Candidates are required to give their answers in their own words

as far as practicable.

1. Answer any ten questions :

- (a) What is Good's buffer?
- (b) Calculate the amount of K₂Cr₂O₇ (M.W. : 294.18) required in grams for the preparation of 250 mL

 $\left(\frac{N}{10}\right)$ aqueous solution of it.

- (c) Mention the limitations of Beer-Lambert's law.
- (d) What is quantum yield? Can it be more than unity?
- (e) Define a primary standard substance with a proper example.
- (f) What is a glass electrode?
- (g) Calculate the wavenumber of an electromagnetic radiation of wavelength 280 nm.
- (h) Why phosphorescence lifetimes are greater than fluorescence lifetimes?
- (i) Calculate the pH of 10^{-9} M HCl.
- (j) What is an NFPA diamond?
- (k) Name one acidic and one alkaline biological buffer.
- 2. Answer any four questions :
 - (a) Between normality and molality, which one is a better scale for measuring the concentration of a solution and why? Define buffer capacity. Comment on its dependence on buffer concentration.

2+1+2

5

- (b) What are intrinsic and extrinsic fluors? Draw a Jablonski diagram depicting only the non-radiative transitions. 2+3
- (c) Derive the Henderson-Hasselbalch equation for an acidic buffer HA and its metal salt MA. Calculate the pH of a buffer containing 0.01 M acetic acid and 0.01 M sodium acetate (pK_a of acetic acid = 4.75). 3+2
- (d) What are Green Fluorescent Proteins (GFP)? How can FRET be used to measure the distance between a donor and an acceptor chromophore? 2+3
- (e) What is the basic principle and instrumentation of uv-visible spectroscopy?

Please Turn Over

2×10

- 3. Answer any four questions :
 - (a) (i) What is the principle of Lowry assay of protein estimation?
 - (ii) How is it different from BCA assay?
 - (iii) What is ELISA? Mention two types of it. 4+2+(2+2)
 - (b) (i) Mention the personal and chemical safety precautions that must be observed in a biochemical laboratory.
 - (ii) Why are ethers stored in air-tight dark containers?
 - (iii) Can a secondary standard solution be used without proper standardization? 4+3+3
 - (c) (i) Differentiate between dynamic and static quenching mechanisms.
 - (ii) Which one of them depends on the viscosity coefficient of the medium?
 - (iii) Write the Stern-Volmer equation mentioning how the Stern-Volmer quenching constant can be measured graphically.
 - (iv) Why is fluorescence more sensitive than absorption spectroscopy? (2+2)+2+2+2
 - (d) (i) What is the full form of BCA assay?
 - (ii) How does a BCA assay detect protein concentration?
 - (iii) What factors affect BCA assay?
 - (iv) What are the importances of virtual laboratory? 2+3+2+3
 - (e) (i) Derive a relation between absorbance and transmittance of a sample.
 - (ii) Which of the above is additive and why?
 - (iii) Determine the molar extinction coefficient of a 1.5×10^{-4} M protein solution which shows an absorbance of 0.60 at 280 nm in a 0.5 cm path-length cuvette.
 - (iv) Why is the detector placed perpendicularly to the source of radiation in a fluorescence spectrophotometer? 3+2+3+2