## GURUDAS COLLEGE

## DEPARTMENT OF BIOCHEMISTRY

UG INTERMEDIATE EXAMINATION, 2020

## B.SC BIOCHEMISTRY HONS. SEMESTER II

PAPER Core Course 4 Enzymes (Semester 2) BCM-A-CC-2-4-TH

## TIME 30 mins

FULL MARKS
$10+25$
Choose correct answer:

1. Anion exchanger is itself
a. Positively charged
b. Negatively charged
c. Neutral
d. None of these
2. At physiological pH Histones are
a. basic protein
b. acidic protein
c. neutral
d. none of these
3. Ribozymes are
a. enzymes which use ribose as substrate
b. enzymes working on
c. RNAs with catalytic activities
d. enzyme-RNA complexes
4. The enzyme where catalysis involves transfer of electrons are named as
a. isomerase
b. transferase
c. oxidoreductase
d. lyase
5. Enormous diversity of proteins is due to
a. sequence of amino acids
b. R-groups of amino acids
c. peptide bonds
d. amino groups of amino acids
6. Disulphide bridge forms between two cysteine residues as a result of
a. oxidation of sulphydral group
b. reduction of sulphydral group
c. amide formation
d. none of these
7. Most abundant protein in the human body is
a. hemoglobin
b. keratin
c. collagen
d. immunoglobin
8. An amino acid has three ionizable groups with pKa 's of $2.0,10.5$ and 3.8. What is the pI of this amino acid?
a. 6.25
b. 9.05
c. 5.43
d. 2.90
9. Which of the statements regarding enzymes is false?
a) Enzymes are proteins that function as catalysts.
b) Enzymes are specific.
c) Enzymes provide activation energy for reactions.
d) Enzyme activity can be regulated.
e) Enzymes may be used many times for a specific reaction.
10. The active site of an enzyme
a) remains rigid and does not change shape.
b) is found at the center of globular enzymes.
c) is complementary to the rest of the molecule.
d) contains amino acids without sidechains.
e) None of the above choices are correct.
11. Which of the following is true about Michaelis-Menten kinetics?
a) Km , the Michaelis constant, is defined as that concentration of substrate at which enzyme is working at maximum velocity
b) It describes single substrate enzymes
c) Km , the Michaelis constant is defined as the dissociation constant of the enzyme-substrate complex
d) It assumes covalent binding occurs between enzyme and substrate
12. When the velocity of enzyme activity is plotted against substrate concentration, which of the following is obtained?
a) Hyperbolic curve
b) Parabola
c) Straight line with positive slope
d) Straight line with negative slope
13. Which of the following statements is true about competitive inhibitors?
a) It is a common type of irreversible inhibition
b) In the presence of a competitive inhibitor, the Michaelis-Menten equation becomes $\mathrm{V}_{0}=\mathrm{V}_{\max }[\mathrm{S}] / \alpha \mathrm{K}_{\mathrm{m}}+[\mathrm{S}]$
c) The apparent Km decreases in the presence of inhibitor by a factor $\alpha$
d) The maximum velocity for the reaction decreases in the presence of a competitive inhibitor
14. The catalytic efficiency of two distinct enzymes can be compared based on which of the following factor?
a) $K_{m}$
b) Product formation
c) Size of the enzymes
d) pH of optimum value
15. What is the general mechanism of an enzyme?
a) It acts by reducing the activation energy
b) It acts by increasing the activation energy
c) It acts by decreasing the pH
d) It acts by increasing the pH
16. The allosteric inhibitor of an enzyme $\qquad$
a) Causes the enzyme to work faster
b) Binds to the active site
c) Participates in feedback regulation
d) Denatures the enzyme.
17. The attachment of phosphoryl groups to specific amino acid residues is catalyzed by
a) Diphteria toxin and cholera toxin
b) Dinitrogenase reductase
c) Protein phosphatases
d) Protein kinases
18.Enzymes having slightly different molecules structure but performing identical activity are
a) Apoenzymes
b) Isoenzymes
c) Holoenzymes
d) Coenzymes
